

## Multiple processes of geochemical evolution for the alkaline rocks of Rio Bonito intrusive complex, Rio de Janeiro State, Brazil: $^{40}\text{Ar}/^{39}\text{Ar}$ and U-Pb ages and Lu-Hf isotopes on zircon and constraints on crustal signature

*Múltiplos processos de evolução geoquímica para as rochas alcalinas do complexo intrusivo de Rio Bonito, Estado do Rio de Janeiro, Brasil: idades  $^{40}\text{Ar}/^{39}\text{Ar}$  e U-Pb e isótopos Lu-Hf em zircão – considerações sobre assinatura crustal*

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### Abstract

This article presents geochemical characteristics of the alkaline rocks of Rio Bonito intrusive complex, State of Rio de Janeiro, Brazil, which is constituted mainly by nepheline syenite. The fractional crystallisation of this magma decreases  $\text{K}_2\text{O}/(\text{Na}_2\text{O} + \text{K}_2\text{O})$  and increases  $(\text{Na} + \text{K})/\text{Al}$ . The  $\text{TiO}_2$ ,  $\text{Fe}_2\text{O}_3^*$ ,  $\text{MgO}$ ,  $\text{CaO}$ , and  $\text{P}_2\text{O}_5$  contents indicate fractionation of titanite, ilmenite, and clinopyroxene or amphibole. The total rare earth elements (REEs) are high, and the REE pattern is linear with negative gradient. The nepheline syenite aplite has low REEs, concave REE pattern, and positive Eu anomaly. The ultrabasic and basic mela-nepheline syenite samples have total REEs and light REEs higher than the felsic alkaline rocks. Therefore, the nepheline syenite magma is not derived directly from the alkaline ultrabasic magma. Laser-spot step-heating  $^{40}\text{Ar}/^{39}\text{Ar}$  ages for biotite and amphibole are  $65.03 \pm 0.70$  and  $65.03 \pm 0.46$ . U-Pb ages LA-ICP-MS for two samples are  $65.49 \pm 0.30$  and  $65.18 \pm 0.30$ . Values of  $\epsilon_{\text{Hf}}$  are negative for both samples, indicating an important crustal component in the evolution of Rio Bonito.

**Keywords:** Nepheline syenite; Rio Bonito; Fractional crystallisation; Crustal assimilation;  $^{40}\text{Ar}/^{39}\text{Ar}$  and U-Pb dating.

### Resumo

Este artigo apresenta características geoquímicas das rochas alcalinas do complexo intrusivo de Rio Bonito, estado do Rio de Janeiro, Brasil, que é constituído principalmente por nefelina sienito. A cristalização fracionada desse magma diminui  $\text{K}_2\text{O}/(\text{Na}_2\text{O} + \text{K}_2\text{O})$  e aumenta  $(\text{Na} + \text{K})/\text{Al}$ . Os conteúdos de  $\text{TiO}_2$ ,  $\text{Fe}_2\text{O}_3^*$ ,  $\text{MgO}$ ,  $\text{CaO}$  e  $\text{P}_2\text{O}_5$  indicam fracionamento de titanita, ilmenita, clinopiroxênio e anfibólito. O total de terras raras é alto, e o padrão é linear com gradiente negativo. O nefelina sienito aplito tem concentrações de terras raras baixas com padrão côncavo e anomalia positiva de Eu. As amostras de mela-nefelina sienito ultrabásico e básico têm total de terras raras e terras raras leves maiores que as rochas alcalinas felsicas. Portanto, o magma nefelina sienítico não é derivado diretamente do magma alcalino ultrabásico. Idades  $^{40}\text{Ar}/^{39}\text{Ar}$  para biotita e anfibólito são  $65,03 \pm 0,70$  e  $65,03 \pm 0,46$ . Idades U-Pb LA-ICP-MS de duas amostras são  $65,49 \pm 0,30$  e  $65,18 \pm 0,30$ . Os valores de  $\epsilon_{\text{Hf}}$  são negativos para ambas as amostras, indicando um importante componente crustal na evolução de Rio Bonito.

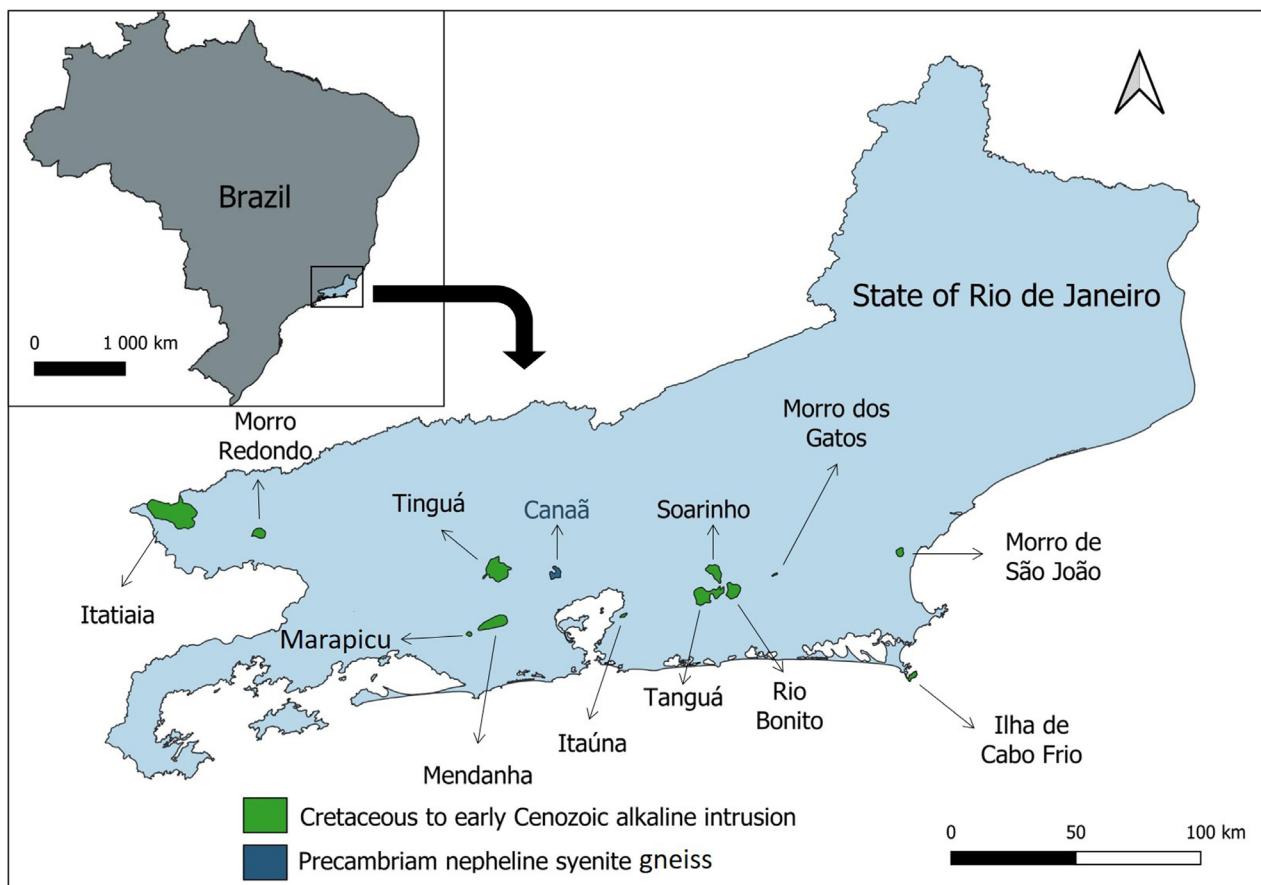
**Palavras-chave:** Nefelina sienito; Rio Bonito; Cristalização fracionada; Assimilação crustal; Datação  $^{40}\text{Ar}/^{39}\text{Ar}$  e U-Pb.

## INTRODUCTION

In the southeastern Brazilian coast region, there is a suit of felsic alkaline intrusive bodies dated from Cretaceous to Early Cenozoic (Figure 1), such as: Itatiaia (Brotzu et al., 1997), Morro Redondo (Brotzu et al., 1989), Tinguá, Mendenha, Marapicu (Motoki et al., 2007a; Silva et al., 2015), Itaúna (Motoki et al., 2008c), Tanguá, Soarinho, Rio Bonito (Motoki et al., 2010), Morro dos Gatos (Motoki et al., 2012a; Geraldes et al., 2013a), Cabo Frio Island (Sichel et al., 2008; Motoki et al., 2013), and Morro de São João (Brotzu et al., 2007; Mota et al., 2009; Geraldes et al., 2013b). These stocks form a WNW-ESE trend called Poços de Caldas-Cabo Frio magmatic alignment (Sadowski and Dias Neto, 1981), and are constituted mainly by nepheline syenite and alkaline syenite. The syenitic stocks are cross-cut by dykes of phonolite, trachyte, and lamprophyre. Some of them have strongly welded and secondary-flowed vent-filling subvolcanic tuff breccia, such as happen in Itatiaia, Mendenha, Itaúna, Morro dos Gatos, and Cabo Frio Island (Motoki et al., 2007c, 2008b; Mota and Geraldes, 2006). The intrusive depth is estimated

to be 3 km based on the geologic mode of occurrence and fission track in apatite (Motoki et al., 2007b, 2008a; Motoki and Sichel, 2006, 2008). This alkaline alignment has been evoked to constraint the influence of Trindade Plume based on chemistry and isotopes signature according to Thompson et al. (1998) and references therein.

According to Motoki et al. (2010, 2013) and Sichel et al. (2012), the chemical variation for the mafic alkaline rocks originate from mafic minerals fractionation, whereas for the felsic alkaline rocks originate from leucite and K-rich alkaline feldspar fractionation. The fractionation of the felsic alkaline magmas decreases  $K_2O/(Na_2O + K_2O)$ . In addition, the above-mentioned authors presented evidences of strong effects of continental crust assimilation based on Silica Saturation Index (SSI). The SSI is a geochemical parameter calculated from all of the major oxides but  $TiO_2$  and  $P_2O_5$  divided by its molecular masses. Rocks with positive SSI are  $SiO_2$ -oversaturated, and those with negative SSI are  $SiO_2$ -undersaturated. The gradual passage from  $SiO_2$ -undersaturated rocks to the  $SiO_2$ -oversaturated ones within the same intrusive body, crossing over the



**Figure 1.** Distribution map for the felsic alkaline bodies of State of Rio de Janeiro. With the exception of Canaã complex, which is Cambrian nepheline syenite gneiss, they are Cretaceous to early Cenozoic intrusive bodies.

thermal divide between nepheline and quartz, is a peculiar phenomenon.

This article presents the results and considers the geochemical evolution of the nepheline syenite magma of Rio Bonito intrusive complex. The authors performed fieldwork, lithologic and petrographic observations, geochemical analyses,  $^{40}\text{Ar}/^{39}\text{Ar}$  laser-spot step-heating datings, U-Pb datings and Lu-Hf analyses for the alkaline rocks of Rio Bonito intrusive complex, situated in the central region of Rio de Janeiro state, Brazil. The new data indicates the geochemical effects of fractional crystallization and continental crust influence based on silica saturation index calculated according to the form proposed by Motoki et al. (2010) and Sichel et al. (2012). U-Pb and Lu-Hf isotopes were performed using laser ablation inductively coupled plasma mass spectrometer (LA-ICP-MS) in zircons and provided crucial information about melt provenance once U-Pb dating performed in zircon grains can give precise crystallization ages whereas epsilon hafnium ( $\epsilon_{\text{Hf}}$ ) records the degree to which a melt incorporates juvenile mantle versus reworked preexisting crust sources (Bataille et al., 2017). We also present here the SSI proposed by Motoki et al. (2010) and Sichel et al. (2012) as a good parameter to indicate continental crust provenance instead of contamination as previously suggested by those authors.

## RIO BONITO INTRUSIVE COMPLEX

The Rio Bonito intrusive complex occurs at  $22^{\circ}40' \text{S}$ ,  $42^{\circ}38' \text{W}$ , about 65 km to the east-northeast of the city of Rio de Janeiro, Brazil. The distribution area of alkaline rocks is about  $7 (\text{NE-SW}) \times 6 \text{ km (NW-SE)}$ , forming an outstanding massif with approximate relative height of 900 m. The Tanguá and Soarinho intrusive complexes occur very close to Rio Bonito intrusive body (Figure 1).

The basement of the alkaline intrusion is composed mainly of Cambrian orthogneiss and paragneiss of the Pan-African continental collision event (e.g. Heilbron and Machado, 2003; Schmitt et al., 2004), with metamorphic ages of 510 to 550 Ma. The metamorphic basement is cut by post-tectonic granite dykes (Valeriano et al., 2011) and silicified tectonic breccia of the late stage of the continental collision event (Motoki et al., 2011, 2012b). They are cut by early Cretaceous mafic dykes intruded by hydraulic shear fracturing (Motoki et al., 2009), which correspond to a part of feeder dykes of the Paraná continental flood basalt (Stewart et al., 1996).

The Rio Bonito complex shows similar geologic mode of occurrence to the other felsic alkaline bodies of the Poços de Caldas-Cabo Frio alignment. The stock is constituted mainly by nepheline syenite and alkali feldspar syenite. Similar to the case of Tanguá complex, trachytic and phonolitic dykes

are scarce, and in the northeastern border, welded subvolcanic tuff breccia is present. The Figure 2 shows sampling site map and indicates the distribution area of felsic alkaline rocks of Rio Bonito complex is much wider than the previous mapping (Heilbron et al., 2016).

## ANALYTICAL TECHNIQUES

### Whole-rock chemistry

The chemical analyses for the alkaline rocks of Rio Bonito intrusive complex have been performed in the Acme Laboratories Ltd., Ontario, Canada, using XRF and ICP-MS (Table 1). The detection limit for the major elements is 0.01% except for MnO and TiO<sub>2</sub> (0.001%); for the trace elements, detection limits range from 0.1 to 30 ppm.

### $^{40}\text{Ar}/^{39}\text{Ar}$ analyses

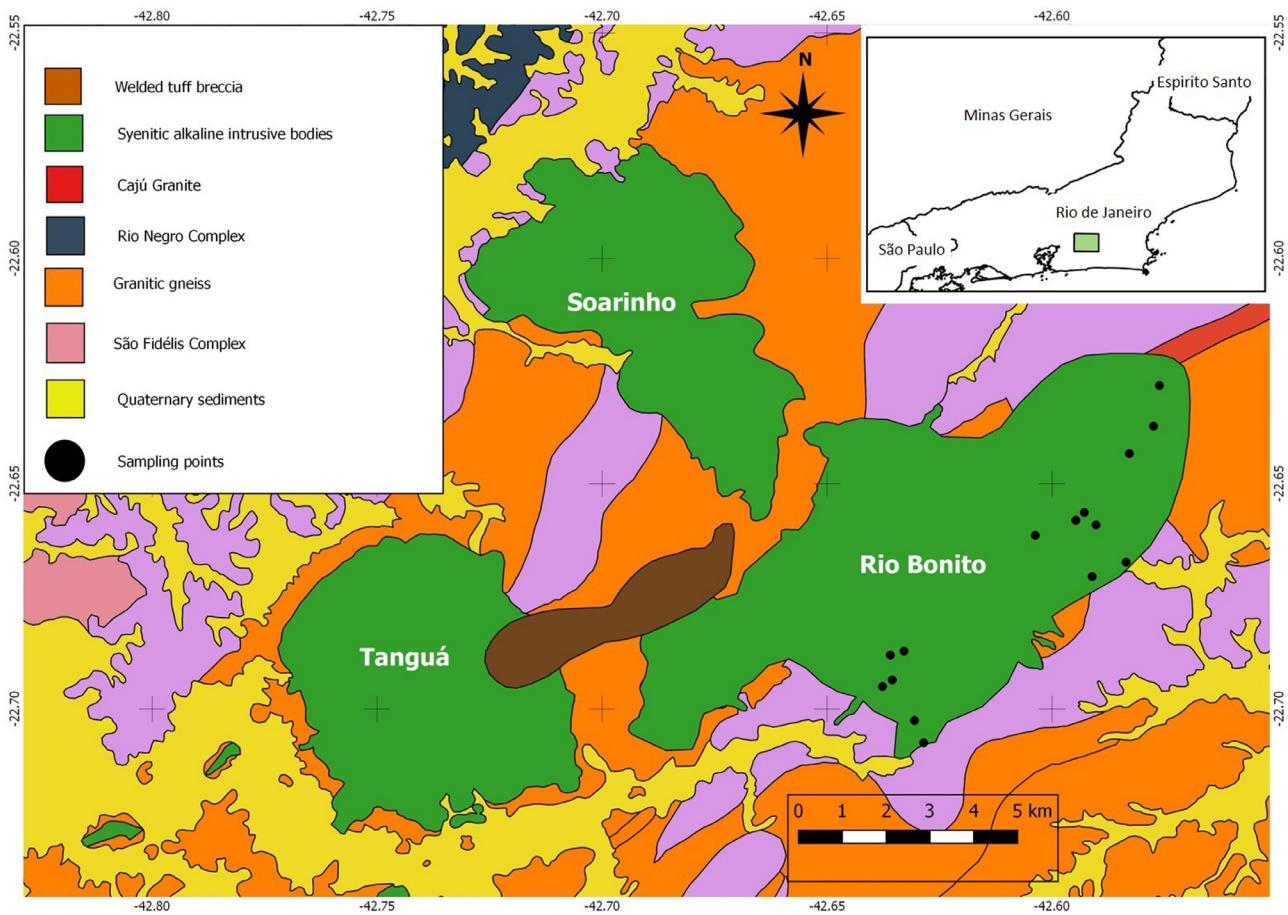
The analytical procedures were performed at Laeter John Centre for Isotope Research, Curtin University (Australia). K and Ar isotope abundances were obtained by mass spectrometry. The procedures include neutron activation to transform  $^{40}\text{K}$  into  $^{39}\text{Ar}$  and step-heating fusion of the whole sample. After the analytical process, two values are obtained for ages:

- plateau age, using criteria of Dalrymple and Lanphere (1974);
- isochron age, with isotopic data extracted from each step of fusion of the same sample.

Correction for atmospheric Ar ( $^{40}\text{Ar}/^{36}\text{Ar} = 295.5$ ) was also performed. The probability of success of the plateau/isochronic ages is done by examining the Pearson's chi-square ( $\chi^2$ ) distribution (Wendt and Carl, 1991). This statistical method uses as input parameters the values of mean squared weighted deviation (MSWD) and the number of steps that define the plateau or the points that define the isochron age.

### U-Pb dating by LA-ICP-MS

Two nepheline syenite samples (Rb-01 and Rb-02) were dated by U-Pb using a LA-ICP-MS. The analyses were made at Isotope Geochemistry Laboratory from Universidade Federal de Ouro Preto, Brazil. Zircon grains from these samples were separated using conventional heavy liquid and magnetic techniques, and the grains were handpicked under a binocular microscope to ensure purity. These zircon grains were mounted onto an epoxy resin disc, polished to about half of the zircon modal grain thickness and coated with gold film. Zircons were documented with Scanning Electron Microscope (SEM) images to reveal their external



**Figure 2.** Geologic map for intrusive complex of Rio Bonito, Tanguá, and Soarinho showing sampling sites.

**Table 1.** New geochemical analyses for the samples of Rio Bonito intrusive complex.  $\text{Fe}_2\text{O}_3^*$  corresponds to total iron as ferric.

	RIB01-A Ns	RIB01-C Ns	RIB01-D Ns	RIB02-B Ns	RIB03 Ns	RIB04 Ns	RIB05-A Ns	RIB05-B Ns	RIB06-A Ns	RIB06-B Ns	RIB07-A Ns
$\text{SiO}_2$	56.24	58.86	56.61	55.9	57.69	56.37	55.31	55.36	57.1	58.51	54.57
$\text{TiO}_2$	1.27	0.45	0.61	1.52	0.78	1.24	1.42	1.44	1.14	0.72	1.39
$\text{Al}_2\text{O}_3$	19.21	20.96	21.37	18.47	18.7	18.66	18.95	18.33	18.64	18.83	18.2
$\text{Fe}_2\text{O}_3^*$	3.89	1.62	2.27	4.88	2.94	3.73	4.73	4.81	3.36	2.65	4.65
$\text{MnO}$	0.19	0.08	0.12	0.18	0.12	0.16	0.18	0.19	0.24	0.17	0.18
$\text{MgO}$	0.86	0.19	0.32	1.22	0.68	0.83	1.02	1.13	0.7	0.38	1.15
$\text{CaO}$	2.76	1.23	1.61	3.36	2.13	2.93	3.19	3.26	2.29	1.76	3.08
$\text{Na}_2\text{O}$	4.27	5.81	6.2	3.98	3.96	4.04	4.86	4.98	4.9	4.49	4.08
$\text{K}_2\text{O}$	8.34	9.05	8.5	8.17	8.8	8.33	7.94	7.62	8.22	8.44	7.88
$\text{P}_2\text{O}_5$	"	0.05	0.1	0.45	0.24	0.31	0.35	0.37	0.09	0.09	0.39
LOI	2.12	1.25	1.62	1.86	1.75	2.35	2.28	1.58	1.55	2.27	2.47
Total	97.32	98.3	97.71	98.13	96.04	96.59	97.95	97.48	96.68	96.05	95.57
Rb	117	118	121	96	99	101	111	98	109	123	104
Sr	2547	601	1429	2036	2160	2047	1199	1269	242	411	1748
Ba	222	34	62	354	503	369	72	91	7	64	274
Zr	301	531	2203	331	216	366	437	354	430	422	369
Y	52	19	30	40	15	35	38	39	57	49	35

Continue...

**Table 1.** Continuation.

	RIB01-A	RIB01-C	RIB01-D	RIB02-B	RIB03	RIB04	RIB05-A	RIB05-B	RIB06-A	RIB06-B	RIB07-A	
	Ns											
Nb	204	94	122	117	59	125	125	131	272	191	118	
Th	9.8	8.7	23.3	7.1	5.4	8.9	8.8	7.6	11.8	12.3	9.1	
U	1.7	2	6.2	1.6	1.3	2.1	1.9	1.7	2.4	3.1	2.2	
Pb	10	10	12	9	9	10	9	9	12	13	10	
Zn	130	60	90	140	90	120	140	140	150	110	150	
Ag	1.1	2.1	9.8	1.2	0.8	1.4	1.8	1.3	1.6	1.7	1.5	
Be	2	2	4	2	2	2	2	2	2	2	2	
Co	2	1	2	2	1	2	2	2	2	2	2	
Mo	8	5	3	5	4	6	5	3	7	11	4	
Sc	1	N.A.	N.A.	3	2	2	2	3	2	1	3	
Cs	0.7	0.5	1.1	0.7	0.6	1	0.8	0.7	0.7	0.9	0.9	
Hf	8.9	8.6	29.6	8.7	4.9	9	10	9.4	12.6	11.1	9.4	
Ta	20.1	7.7	9.4	9.6	4.1	9.5	9.9	10.1	24.7	17.8	9.3	
Ga	23	22	27	22	19	22	23	22	25	24	23	
La	204	98	133	166	93.9	156	162	165	250	194	155	
Ce	428	184	250	326	158	297	314	317	517	378	298	
Pr	52.5	19.9	27.4	42.1	18.1	37.5	39.6	39.8	64.4	45	37.7	
Nd	183	63	90.2	160	63.4	137	146	148	213	149	141	
Sm	27.4	8.4	12.4	24.9	9.3	21.3	23.1	23.4	28.6	20.4	22.3	
Eu	8.53	2.81	4.43	9.82	5.99	8.61	8.7	9.04	6.48	5.51	8.69	
Gd	17.2	5.1	7.8	15.6	5.8	13	14.3	14.7	16.3	13.2	13.7	
Tb	2.3	0.7	1.1	2	0.7	1.7	1.8	1.8	2.4	1.8	1.8	
Dy	11.3	3.8	5.6	9.4	3.5	8.2	9	8.9	12.3	9.8	8.6	
Ho	1.9	0.7	1.1	1.5	0.6	1.3	1.5	1.5	2.2	1.8	1.5	
Er	5	1.9	3.1	3.9	1.6	3.5	3.8	3.8	5.9	5	3.6	
Tm	0.6	0.25	0.45	0.47	0.2	0.42	0.47	0.46	0.74	0.68	0.44	
Yb	3.2	1.5	3.2	2.5	1.1	2.3	2.6	2.6	3.8	4	2.4	
Lu	0.41	0.23	0.55	0.34	0.15	0.32	0.35	0.35	0.49	0.58	0.32	
	RIB08-C	RIB13	RIB14-A	RIB14-B	RIB15-A	RIB15-B	RB-6	Rb-5	Rb-7	Rb-8	R-4	Rb-3
	Ns	Ns	Ns	Ns	Ns	Ns	Ap	Ap	Wt	Wt	Ba	Ub
SiO <sub>2</sub>	54.35	57.23	56.82	58.8	58.36	57.8	56.37	55.13	58.77	58.77	47.68	44.22
TiO <sub>2</sub>	0.32	1.02	1.11	0.94	0.82	0.77	0.16	0.16	0.58	0.59	2.86	3.38
Al <sub>2</sub> O <sub>3</sub>	22.12	19.22	19.28	18.86	18.94	18.36	22.36	23.39	19.46	20	16.89	15.52
Fe <sub>2</sub> O <sub>3</sub>	1.65	3.53	3.73	3.02	2.85	2.82	1.33	0.99	3.02	3.01	9.87	11.77
MnO	0.09	0.17	0.18	0.21	0.19	0.18	0.09	0.06	0.03	0.03	0.6	1
MgO	0.15	0.66	0.69	0.53	0.44	0.43	0.05	0.05	0.21	0.2	2.15	1.6
CaO	1.25	2.23	2.35	1.98	2.02	1.79	0.88	0.97	0.44	0.4	5.58	8.32
Na <sub>2</sub> O	7.61	5.16	5.2	4.44	5.36	5.35	6.78	7.67	4.09	4.15	4.61	7.58
K <sub>2</sub> O	7.51	8.25	8.05	8.8	7.8	8.02	8.35	7.79	6.99	7.22	5.93	3.36
P <sub>2</sub> O <sub>5</sub>	0.04	0.21	0.21	0.11	0.09	0.08	0.02	0.02	0.14	0.14	0.71	0.48
LOI	2.72	2.37	2.33	4.57	4.86	3.71	2.28	2.29	2.55	2.05	3.27	3.01
Total	95.09	97.68	97.61	97.69	96.87	95.59	96.38	96.23	93.74	94.51	96.89	97.22
Rb	147	130	121	122	150	144	133	131	160	165	98	113
Sr	407	811	695	430	226	266	176	161	93	92	785	146
Ba	63	59	42	60	17	28	7	12	44	46	112	20
Zr	274	572	594	555	610	433	860	827	1109	1189	1026	2662
Y	15	31	33	50	36	31	6	5	53	69	127	154
Nb	131	131	129	227	178	152	69	53	199	190	528	N.A.
Th	13.2	10.2	11.4	14.4	21	17.1	32.1	21.5	19	20.8	26.1	159
U	2	2	2.8	2.7	5.4	3.8	10.9	6.8	4.7	5.3	5.6	17.6
Pb	11	12	11	11	18	15	20	13	19	22	11	10

Continue...

**Table 1.** Continuation.

	RIB08-C	RIB13	RIB14-A	RIB14-B	RIB15-A	RIB15-B	RB-6	Rb-5	Rb-7	Rb-8	R-4	Rb-3
	Ns	Ns	Ns	Ns	Ns	Ns	Ap	Ap	Wt	Wt	Ba	Ub
Zn	60	110	110	120	110	100	80	60	110	80	400	500
Ag	1.1	2.3	2.3	2.2	2.6	1.6	4	3.4	4.4	5.1	4.5	12
Be	3	2	3	2	5	3	10	6	3	3	4	8
Co	1	2	2	2	2	2	1	1	2	2	3	5
Mo	5	4	3	12	3	2	N.A.	N.A.	N.A.	N.A.	12	N.A.
Sc	N.A.	2	2	1	1	1	N.A.	N.A.	5	5	5	4
Cs	1.3	0.5	0.9	0.7	1.7	1.9	3.7	2.1	1.5	1.3	1	1.5
Hf	6.4	11.6	12	12.7	14.8	10.9	17.1	13.2	22.5	23.3	28.6	63
Ta	10.1	8.5	9.4	21.9	12.1	10.3	1.5	1.6	12.6	13.1	49.8	98.6
Ga	26	25	24	23	26	24	33	31	29	30	36	47
La	115	147	146	216	172	153	112	67.1	251	278	508	1280
Ce	193	263	273	457	293	262	93	63.2	494	559	1020	2020
Pr	17.4	28.7	31	53.8	30.5	26.8	4.9	3.86	52.3	58.4	127	186
Nd	46.7	100	110	177	101	85.1	10.4	8.5	171	201	435	487
Sm	5.4	15.3	17	23.8	14.9	12	1	0.8	25.1	30.2	62.5	55.8
Eu	1.35	5.93	6.31	5.84	3.85	3.31	0.6	0.53	4.81	5.82	14.7	11.5
Gd	3.4	10.1	11.1	14.6	10	8.2	1	0.5	16.2	21.6	39.4	37.5
Tb	0.5	1.3	1.4	2	1.4	1.1	0.1	N.A.	2.2	2.9	5.3	5.1
Dy	2.7	6.3	7	10.4	7.3	6.1	0.7	0.6	11.2	13.9	26.6	27.8
Ho	0.5	1.1	1.2	1.9	1.3	1.1	0.2	0.1	2	2.5	4.7	5.3
Er	1.5	3.1	3.3	5.1	3.6	3.1	0.7	0.5	5.4	6.6	12.6	15.9
Tm	0.22	0.41	0.42	0.64	0.49	0.41	0.14	0.12	0.74	0.89	1.59	2.22
Yb	1.3	2.5	2.5	3.5	3.1	2.6	1.2	1	4.6	5.4	8.7	13.4
Lu	0.19	0.36	0.37	0.48	0.46	0.38	0.24	0.2	0.71	0.81	1.21	2.09

Ns: nepheline syenite and nepheline-bearing alkaline syenite; Ap: nepheline syenitic aplite; Wt: welded lapilli tuff; Ub: ultrabasic mela-nepheline syenite; Ba: basic mela-nepheline syenite.

and internal structures. The data were selected to ensure discordance less than 5%, and low common lead. The data are presented in Tera and Wasserburg (1972) Concordia constructed by using Isoplot program (Ludwig, 2003).

#### *Lu-Hf isotopes in zircons*

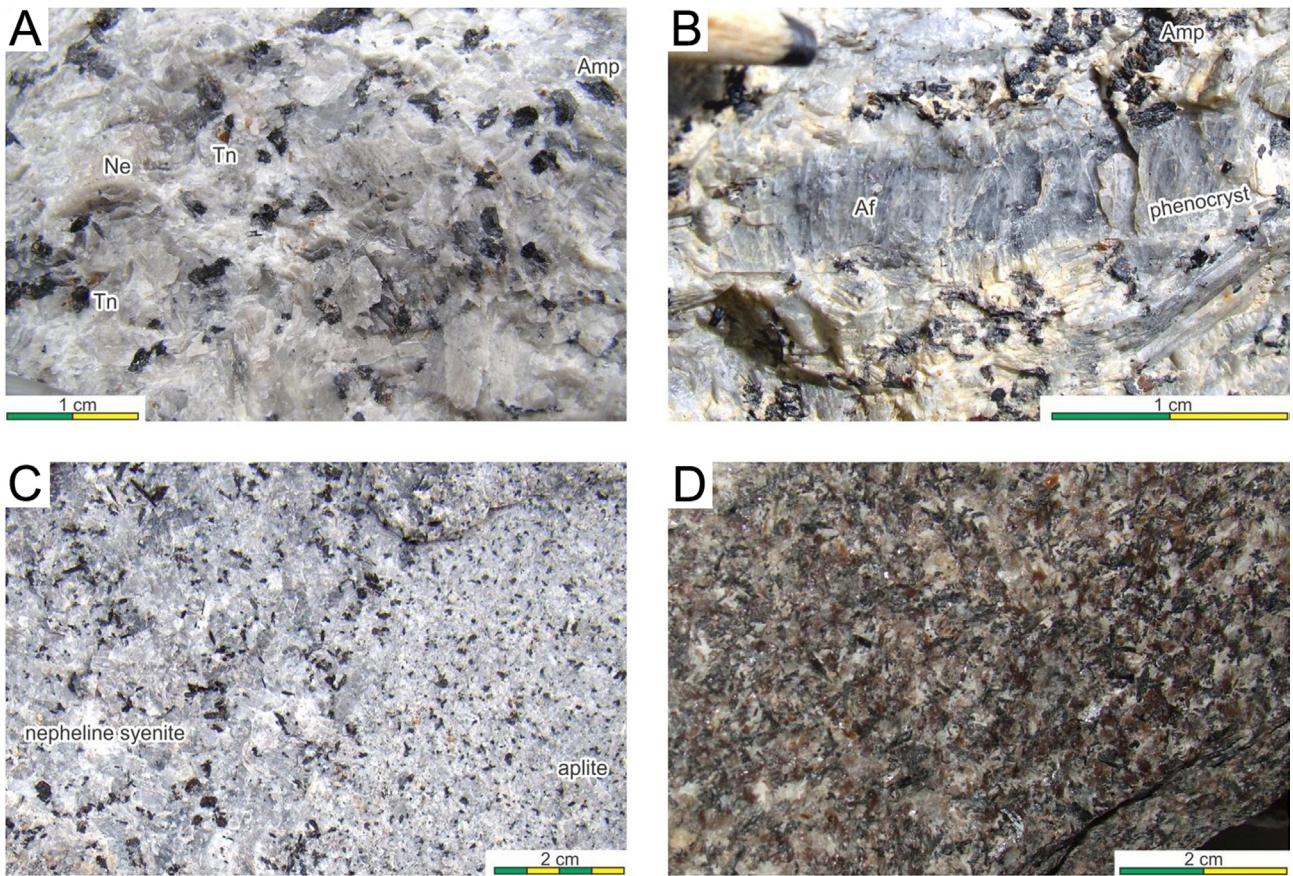
Zircon Lu-Hf isotopes were obtained from the original locations of the LA-ICP-MS analyses. Zircons constrain important tool analysing those heavy isotopes once they do not fractionated from each other during magmatic processes and can imprint the melts signature from which they crystallized. Those analyses were made at Isotope Geochemistry Laboratory from Universidade Federal de Ouro Preto, Brazil.

#### **Lithologic and petrographic characteristics**

Most of the constituent rocks of Rio Bonito complex are nepheline syenite and nepheline-bearing alkali feldspar syenite (Figure 3A). In general, no mineral orientation and cumulative texture are observed. Some samples contain alkaline feldspar phenocrysts with typical size of 2.5 × 1 cm

(Figure 3B). About 80% in volume of the constituting minerals are intermediate alkaline feldspar showing micro-interlocking perthite. Nepheline is slightly pinkish and occurs in the interstitial space of alkaline feldspar framework and generally altered into analcime. Mafic mineral is amphibole with subordinate amount of biotite. The presence of titanite is notable (Figure 4A). Apatite and zircon occur as accessory minerals. Veins of nepheline syenite aplite of 10 to 20 cm in width are commonly found. The grain size of the aplite ranges from 0.1 (Figure 4B) to 1 mm (Figure 3C).

There are two samples of coarse-grained rocks with abundant mafic minerals (Figure 3D) which are constituted by alkaline feldspar, nepheline, clinopyroxene, amphibole, and titanite (Figure 4C). The clinopyroxene is euhedral showing zoning and nepheline is generally altered into analcime. These rocks have mineralogical composition of nepheline syenite and, due to its high amount of mafic minerals and chemical composition, in this paper, the authors classify samples Rb-3 and Rb-4, respectively, ultrabasic mela-nepheline syenite and basic mela-nepheline syenite. Such classification is in accordance with IUGS recommended nomenclature for igneous rocks (Le Maitre, 2002). Similar alkaline



Af: alkaline feldspar; Amp: amphibole; Ne: nepheline; Tn: titanite.

**Figure 3.** Akaline rocks of Rio Bonito complex. (A) Nepheline syenite, RIB-01; (B) nepheline syenite with alkaline feldspar phenocryst, RIB-13; (C) nepheline syenite aplite, Rb-6; (D) ultrabasic mela-nepheline syenite, Rb-3.

mafic rocks are found as enclaves captured by nepheline syenite in the intrusive bodies of Morro de São João, Cabo Frio Island, and São Sebastião Island.

A gravel of welded lapilli tuff of  $30 \times 15$  cm in size is found in a stream bottom. This rock contains angular fragments of trachyte and syenite smaller than 2 cm (Figure 4D). The mineral grains smaller than 0.1 mm have highly fragmented shape. The matrix texture indicates high-grade welding and strong secondary flow. Because of the advanced plastic deformation, flattened lens-shaped essential fragments are no more observable. These textural characteristics are common in vent-filling subvolcanic welded tuff breccia, observed in the alkaline intrusive complexes of Morro Redondo, Mendenha, Itaúna, Tanguá, Morro dos Gatos, and Cabo Frio Island (Brotzu et al., 1989; Motoki et al., 2007b, 2007c, 2008b, 2008c; Sichel et al., 2008). The limited field occurrence suggests that the welded tuff breccia could be originated from a pyroclastic dyke as seen in Mendenha and Morro dos Gatos complexes (Motoki et al., 2007a, 2012a). Samples RIB-01, RIB-07, and RIB-08 of Rio Bonito are

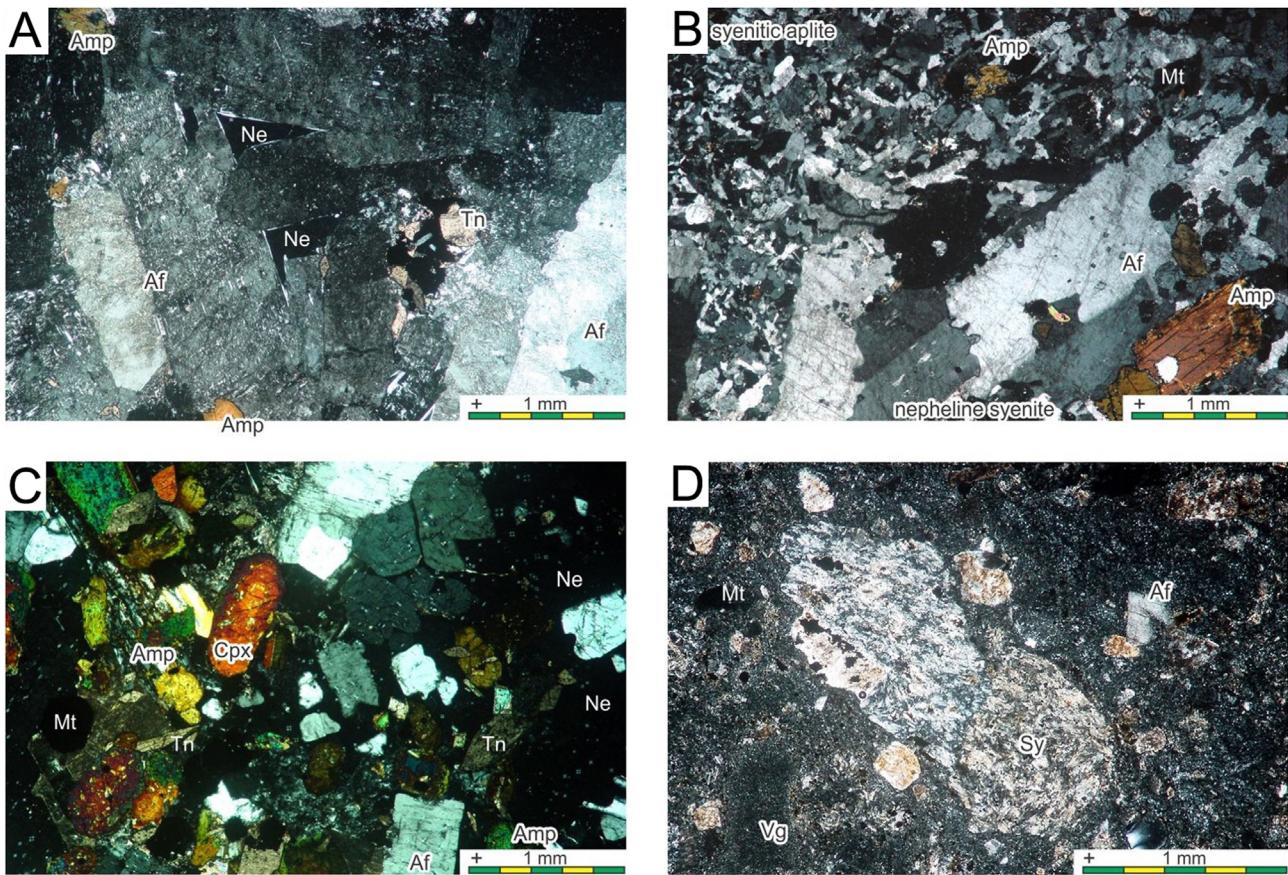
constituent of trachyte dyke and trachytic subvolcanic pyroclastic conduit and fissure.

### Major and trace elements

The classification diagram of Le Bas et al. (1986) shows that all of the analysed samples plot on the areas of alkaline rocks (Figure 5). Most of the rocks are felsic and their  $\text{SiO}_2$  contents range from 52.43 to 61.72 wt% (Table 1).

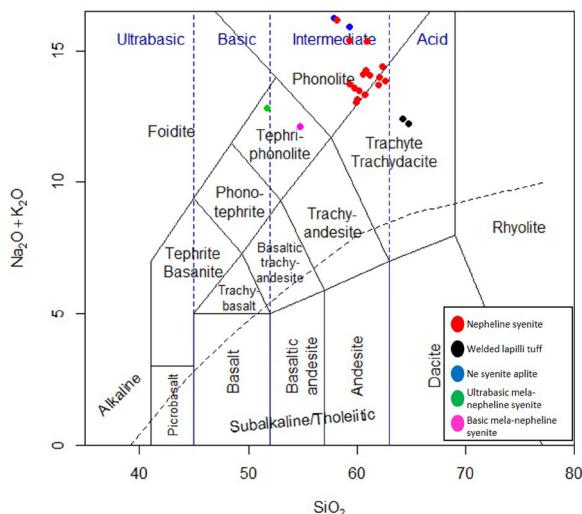
The ultrabasic and basic mela-nepheline syenite of the Rio Bonito complex have differentiation index (D.I. Thornton and Tuttle, 1960) respectively of 69.23 and 55.47 wt%, which are significantly lower than the felsic alkaline rocks, with average value of 89.93 wt%. They have high normative nepheline contents respectively 15 and 30 wt%, being plotted on the tephri-phonolite field (Figure 5). The ultrabasic mela-nepheline syenite has exceptionally 2.66 wt% of normative acmite, although the other rocks of Rio Bonito have none.

The nepheline syenite aplite have high normative nepheline content constitution, respectively 20.22 and 25.41 wt%



Af: alkaline feldspar; Amp: amphibole; Ne: nepheline altered into analcite; Cpx: clinopyroxene; Mt: magnetite; Tn: titanite; Sy: syenitic rock (accidental fragment).

**Figure 4.** Photomicrography of the alkaline rocks of Rio Bonito intrusive complex. (A) Nepheline syenite, RIB-13; (B) syenitic aplite, Rb-6; (C) ultrabasic mela-nepheline syenite, Rb-3; (D) welded lapilli tuff, RIB-08. The symbol + means crossed polarizer.



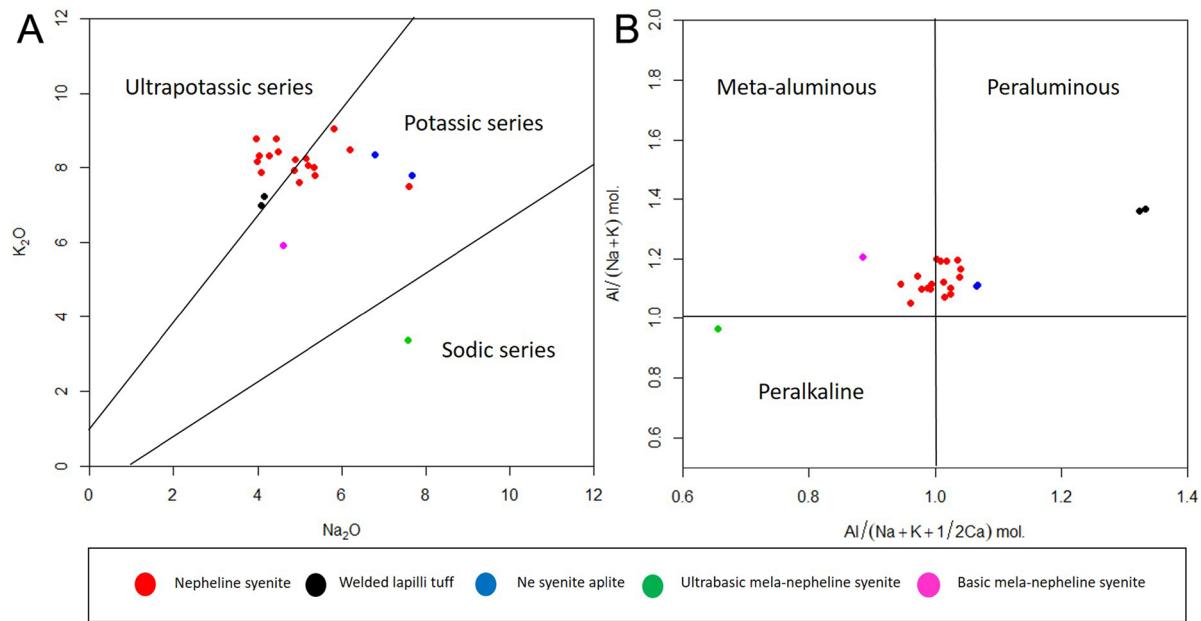
**Figure 5.**  $\text{Na}_2\text{O} + \text{K}_2\text{O}$  vs.  $\text{SiO}_2$  (wt%) diagram for the alkaline rocks of Rio Bonito complex. The classification nomenclature is after Le Bas et al. (1986). Limit between alkaline and non-alkaline rocks defined by Irvine and Baragar (1971).

and normative corundum of 0.62 wt%. These rocks are potassic, with respective  $\text{K}_2\text{O}/(\text{K}_2\text{O} + \text{Na}_2\text{O})$  ratio of 0.50 and 0.55. No normative clinopyroxene is found. Their differentiation indexes are notably high, 94.74 and 94.47 wt%, respectively.

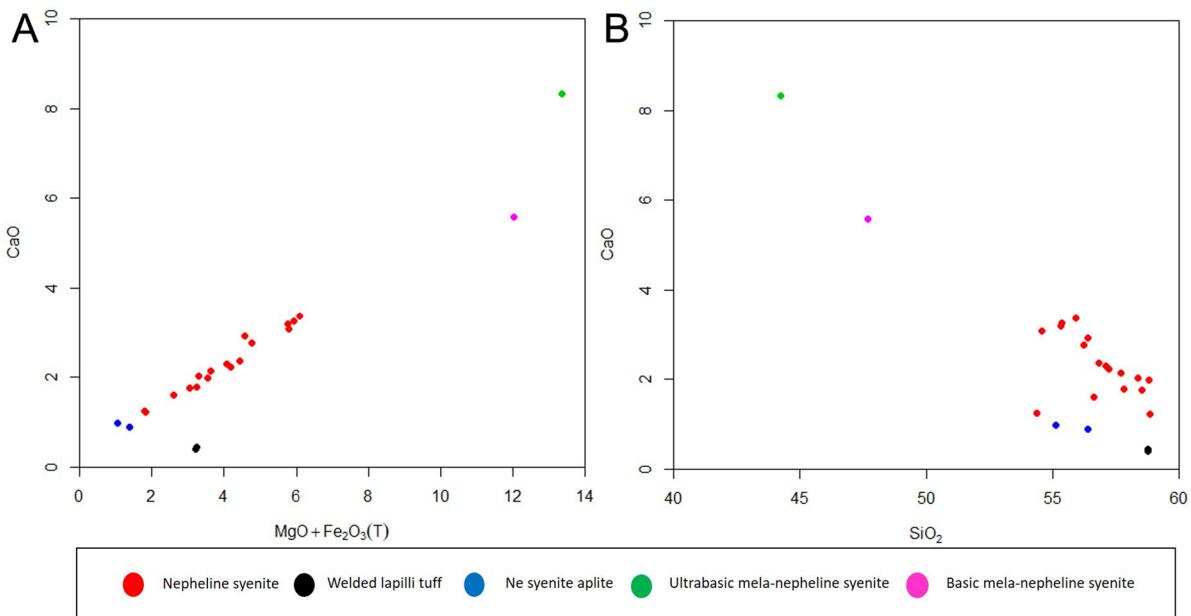
The diagram of  $\text{Na}_2\text{O}$  vs.  $\text{K}_2\text{O}$  shows that the felsic alkaline rocks have high total alkalis and the  $\text{K}_2\text{O}$  is higher than  $\text{Na}_2\text{O}$  (Figure 6A). Most of the rocks of Rio Bonito complex are ultrapotassic or potassic (Figure 6A).

The alkali-alumina diagram (Figure 6B) shows that highly fractionated felsic rocks tend to be peralkaline and most of felsic alkaline rocks of Rio Bonito complex are less fractionated and have no normative acmite. Trachytic rocks of Rio Bonito are peraluminous. The ultrabasic and basic mela-nepheline syenite are peralkaline and metaluminous, respectively.

The good linear convergence on the diagrams of  $\text{MgO} + \text{Fe}_2\text{O}_3^*$  vs.  $\text{CaO}$  and  $\text{SiO}_2$  vs.  $\text{CaO}$ , with respective  $R^2$  of 0.895 and 0.629, is attributed to fractionation of clinopyroxene and amphibole (Figure 7).



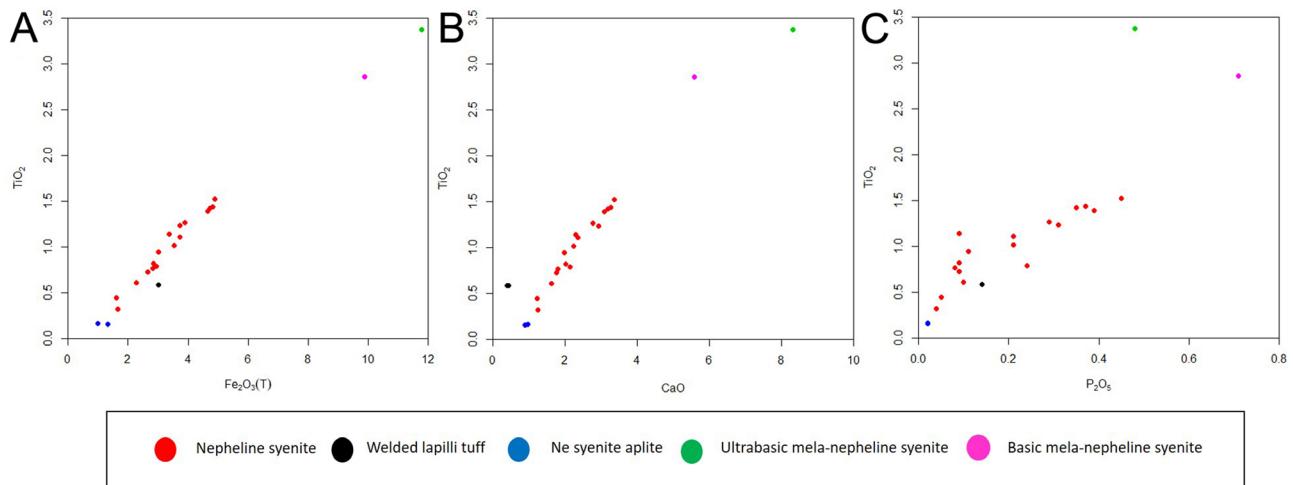
**Figure 6.** Geochemical classification diagrams for the alkaline rocks of Rio Bonito intrusive complex. (A)  $\text{Na}_2\text{O}$  vs.  $\text{K}_2\text{O}$  wt% according to Middlemost (1975); (B) Alkali-alumina saturation diagram after Maniar and Piccoli (1989).



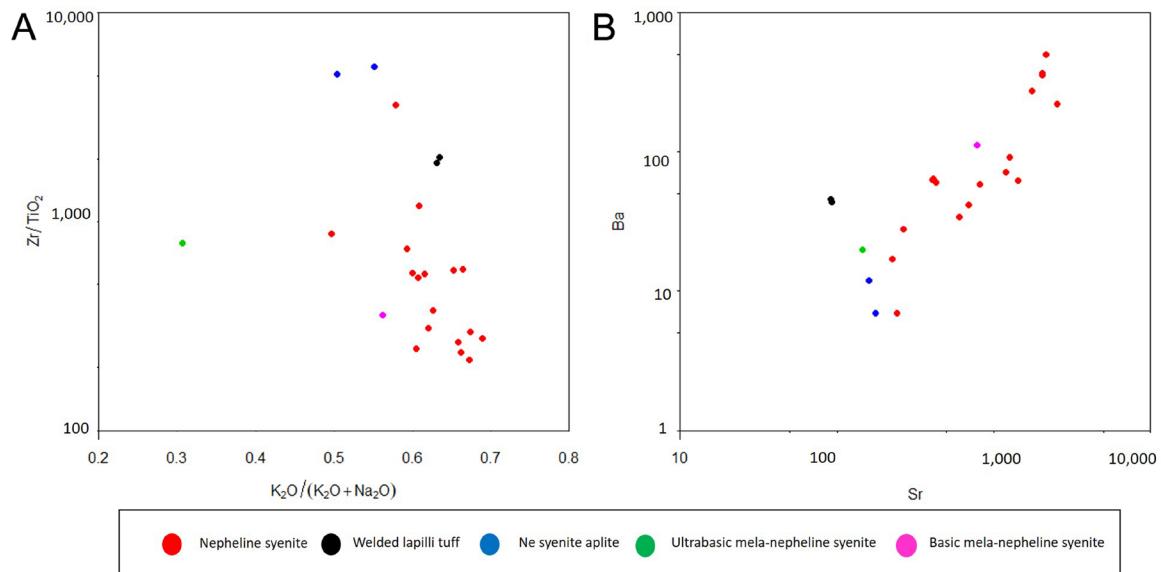
**Figure 7.** Geochemical effects of mafic mineral fractionation represented by  $\text{MgO} + \text{Fe}_2\text{O}_3^*$  vs.  $\text{CaO}$  wt%; (B)  $\text{SiO}_2$  wt% vs.  $\text{CaO}$  wt%.

The  $\text{TiO}_2$  vs.  $\text{Fe}_2\text{O}_3^*$  diagram exhibits well-converged positive correlation with  $R^2 = 0.839$ , suggesting strongly ilmenite fractionation (Figure 8A). The positive correlations on the diagrams of  $\text{TiO}_2$  vs.  $\text{CaO}$  and  $\text{TiO}_2$  vs.  $\text{P}_2\text{O}_5$ , with respective  $R^2$  of 0.924 and 0.635, indicate titanite fractionation (Figures 8B and 8C), which is common in these alkaline rocks.

The large ion lithophile elements (LILE) ratios cannot express well the fractional crystallization processes of the studied rocks. The wide range of the  $\text{Rb/Ba}$  and  $\text{Rb/Sr}$  ratios suggests a strong signature of continental crust in the intrusive bodies. The concentrations of Ba and Sr have a positive correlation (Figure 9).



**Figure 8.** Variation diagrams in relation to titanium. (A)  $\text{Fe}_2\text{O}_3^*$  vs.  $\text{TiO}_2$  wt%; (B)  $\text{CaO}$  vs.  $\text{TiO}_2$  wt%; (C)  $\text{P}_2\text{O}_5$  vs.  $\text{TiO}_2$  wt%.



**Figure 9.** Variation diagrams for Zr, Sr, and Ba showing fractional crystallization process. (A)  $\text{K}_2\text{O}/(\text{K}_2\text{O} + \text{Na}_2\text{O})$  vs.  $\text{Zr}/\text{TiO}_2$ ; (B) Sr vs. Ba.

The Figure 10 shows C1 chondrite normalized rare earth elements (REE) of the alkaline rocks of Rio Bonito. The average total REEs of the felsic rocks of Rio Bonito complex is 646 ppm, which is similar to that of Tanguá complex, 703 ppm, and about twice of Morro de São João, 333 ppm. The REE patterns are characterised by enrichment of light REEs. The average  $\text{La}_{\text{N}}/\text{Yb}_{\text{N}}$  (CI chondrite normalized values) of the felsic rocks of Rio Bonito complex is 43.6, which is similar to Tanguá, 49.1, and almost double of Morro de São João,

29.1. The average  $\text{La}_{\text{N}}/\text{Sm}_{\text{N}}$  of the felsic rocks of Rio Bonito, Tanguá, and Morro de São João is, respectively, 9.63, 6.34, and 11.28. Most of the samples have almost linear REE variation pattern from La to Yb. However, the nepheline syenite aplites of Rio Bonito show concave-up pattern with low total REEs.

On the other hand, total REEs concentration of the ultrabasic and basic mela-nepheline syenite of Rio Bonito complex, respectively 2,267 and 4,150 ppm, is much higher than that of the felsic alkaline rocks. They are about 4 times of

alkaline ultrabasic and basic rocks of Morro de São João body, 837 ppm in average.

The Figure 11 shows primitive mantle normalised multi-elements spider diagrams comparing LILE and high field-strength elements (HFSE). The Sr, Ba, and P shows strong negative anomalies, being 10 to 100 times lower than the adjacent elements. The negative anomaly variation of Sr and Ba is of wide range. Some felsic rocks of Rio Bonito show notable negative anomaly of Sm and Y.

## Geochronology

### $^{40}\text{Ar}/^{39}\text{Ar}$ laser-spot dating

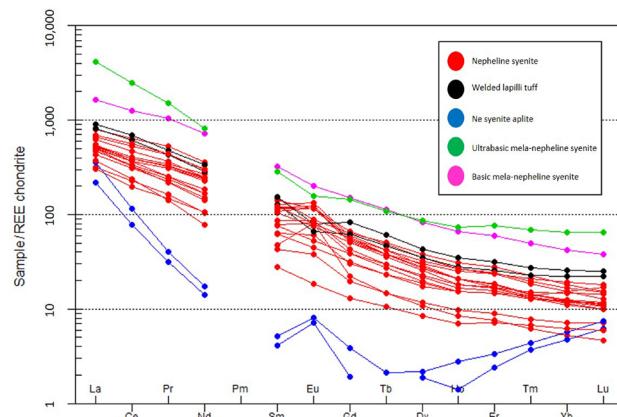
Two nepheline syenite samples (RIB01-B and RIB01-A) of Rio Bonito complex have been submitted to laser-spot step-heating  $^{40}\text{Ar}/^{39}\text{Ar}$  dating. The samples selected for the dating

are highly  $\text{SiO}_2$ -undersaturated rock to avoid xenocrysts originated from the wall rock. The dating for biotite of the sample RIB01-B shows a coherent plateau of  $65.03 \pm 0.70$  Ma with MSWD of 0.48 (Figure 12A). The amphibole of the sample RIB01-A demonstrates a plateau of  $65.03 \pm 0.46$  Ma with MSWD of 0.09 (Figure 12B). The coherent plateaus endure high accuracy of the datings.

These  $^{40}\text{Ar}/^{39}\text{Ar}$  ages are close to the Rb-Sr whole rock age of adjacent intrusive complex of Tanguá, 66.8 Ma (Motoki et al., 2010). These ages are according to the general tendency of alkaline intrusive body of Poços de Caldas-Cabo Frio magmatic alignment whose ages become younger from west to east.

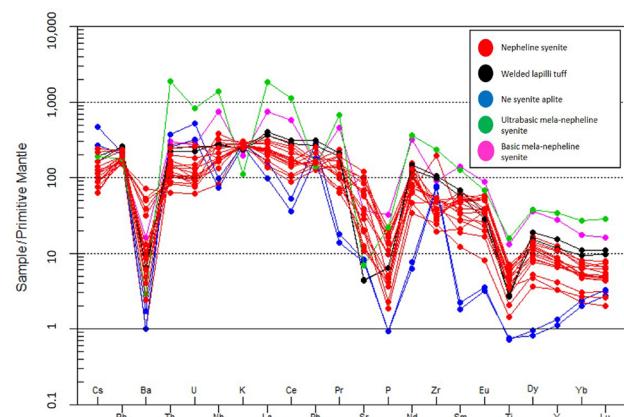
### $U\text{-Pb}$ and $\text{Lu}\text{-Hf}$ performed in zircon grains

Twenty-eight laser spots were made for each sample. Sample RB-01 shows a concordant age of  $65.47 \pm 0.6$  Ma

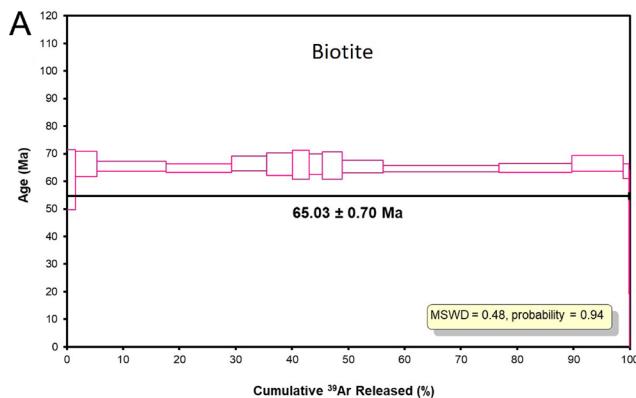


REE: rare earth elements.

**Figure 10.** Normalised rare earth elements for the alkaline rocks of Rio Bonito complex. The CI chondrite data are from Boynton (1984).

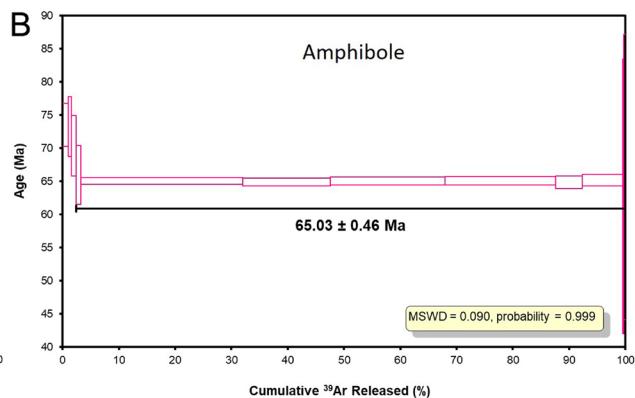


**Figure 11.** Primitive mantle normalised multi-elements spider diagrams contrasting large ion lithophile elements and high field-strength elements. The data of primitive mantle are after Sun and McDonough (1989).



MSWD: mean squared weighted deviation.

**Figure 12.**  $^{40}\text{Ar}-^{39}\text{Ar}$  step-heating dating diagrams for nepheline syenite samples of Rio Bonito complex, State of Rio de Janeiro, Brazil: (A) biotite of the sample RIB01-B; (B) amphibole of the sample RIB01-A.



with  $\text{MSWD} = 0.49$  and probability 48% according to the Figure 13A. The selected spots have discordance of 0% and only one point that have 4% discordance (see Appendix 1). The sample RB-02 shows a concordant age of  $65.18 \pm 0.6$  Ma with  $\text{MSWD} = 0.013$  and probability of 91% according to the Figure 13B. The selected spots have discordance of between 0% (see Appendix 2).

Twenty-four and twenty Lu-Hf analyses were made for the samples RB-01 and RB-02 respectively. Sample RB-01 exhibits variable initial  $^{176}\text{Hf}/^{177}\text{Hf}$  ratios from 0.282102 to 0.282586 (Figure 14A) and  $\epsilon_{\text{Hf}}$  values ranging from -5.6 to -25.7 (Figure 14B) with  $T_{\text{DM}}$  values between 1.1 and 2.4 Ga (see Appendix 3). Sample RB-02 exhibits variable initial  $^{176}\text{Hf}/^{177}\text{Hf}$  ratios from 0.281816 to 0.282559 (Figure 14C) and  $\epsilon_{\text{Hf}}$  ranging from -6.5 to -33.2 (Figure 14D) with  $T_{\text{DM}}$  values between 1.2 and 2.8 Ga (see Appendix 4).

## DISCUSSION

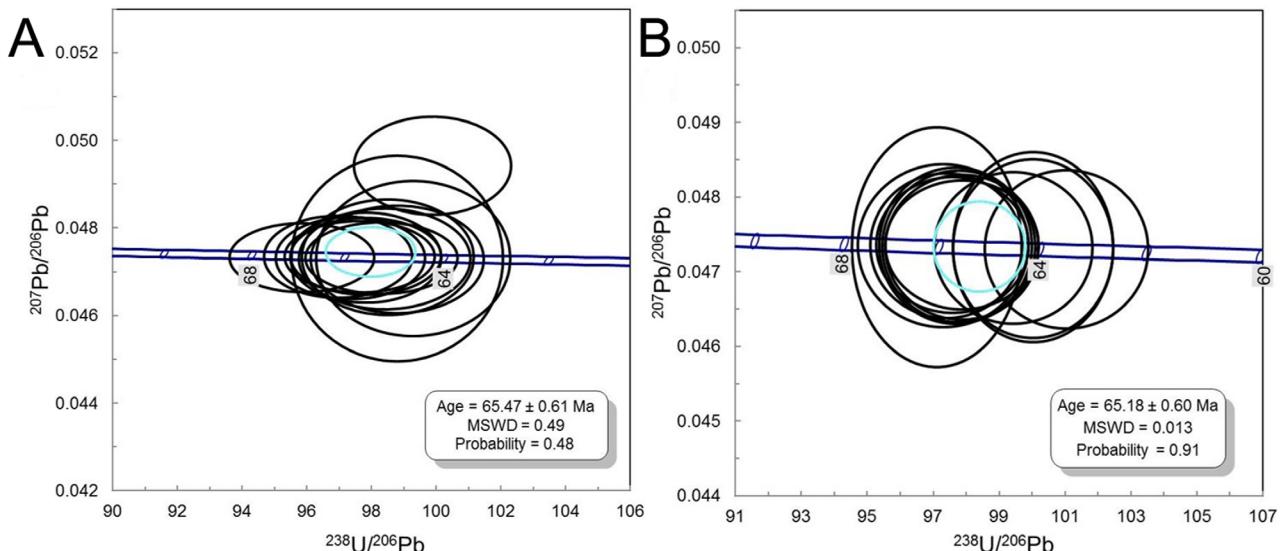
Silica undersaturated magmas cannot evolve to silica oversaturated ones by fractional crystallization, and crustal assimilation is evoked to explain this phenomenon (Sichel et al., 2012). The continental crust assimilation transforms  $\text{SiO}_2$ -undersaturated original nepheline syenite magma into  $\text{SiO}_2$ -oversaturated alkali feldspar syenite magma, crossing over the thermal divide between nepheline and quartz. The SSI as originally proposed represents the grade of the crust assimilation (Motoki et al., 2010). The rocks with negative SSI are  $\text{SiO}_2$ -undersaturated and have Norm nepheline.

Almost all of the felsic rocks with positive SSI, especially alkali feldspar syenite, are  $\text{SiO}_2$ -oversaturated.

In addition to the nepheline-quartz incompatibility, there is another thermodynamic incompatibility. It is based on alkali-alumina saturation. The rocks with  $\text{Na} + \text{K} > \text{Al}$  mol. are classified as peralkaline. They can crystallise alkaline clinopyroxene and alkaline amphibole, such as aegirine, aegirine-augite, riebeckite, and barkevite. On the other hand, highly alumina-rich rocks with  $\text{Na} + \text{K} + 1/2\text{Ca} < \text{Al}$  mol. are classified as peraluminous. They can crystallise muscovite. Most of the igneous rocks have  $\text{Al}/(\text{Na} + \text{K}) > 1$  and  $\text{Na} + \text{K} + 1/2\text{Ca} > \text{Al}$ , being classified as meta-aluminous. Peralkaline magmas cannot transform into peraluminous ones by fractional crystallisation (Motoki et al., 2010). It is similar to the relation between  $\text{SiO}_2$ -undersaturated and oversaturated magmas.

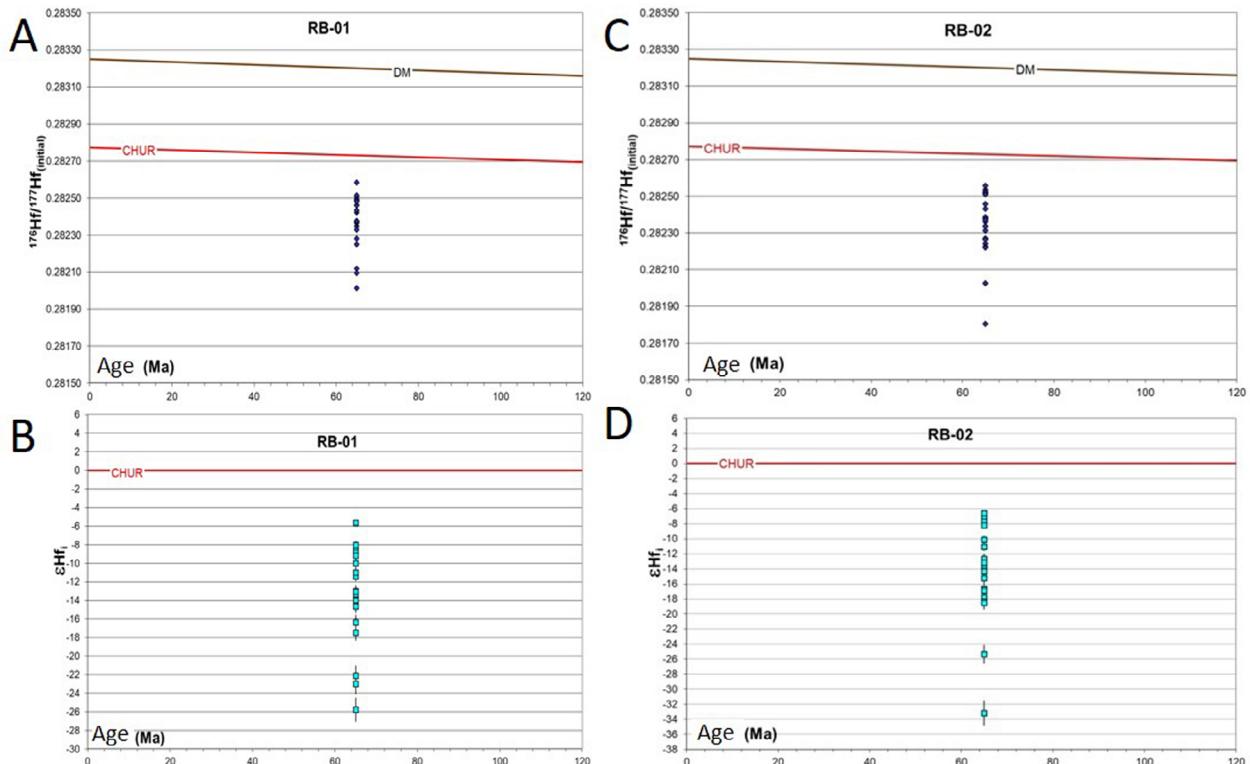
The Figure 15A shows that the alkaline rocks of Rio Bonito constitute a continuous trend from  $\text{SiO}_2$ -undersaturated to oversaturated compositions and from peralkaline to peraluminous one, crossing over the above-mentioned two thermodynamic incompatibilities. This phenomenon can be explained by the crustal assimilation. The fractional crystallisation trend is of the same direction and the opposite sense. Figure 15B shows that fractional crystallisation of the felsic alkaline magmas decreases either SSI and  $\text{K}_2\text{O}/(\text{K}_2\text{O} + \text{Na}_2\text{O})$ , forming a positive correlation trend. On the other hand, the continental crust assimilation increases SSI, showing vertical trends.

Because of the continental crust assimilation effects, the trachytic rocks of Rio Bonito and Tanguá massifs have

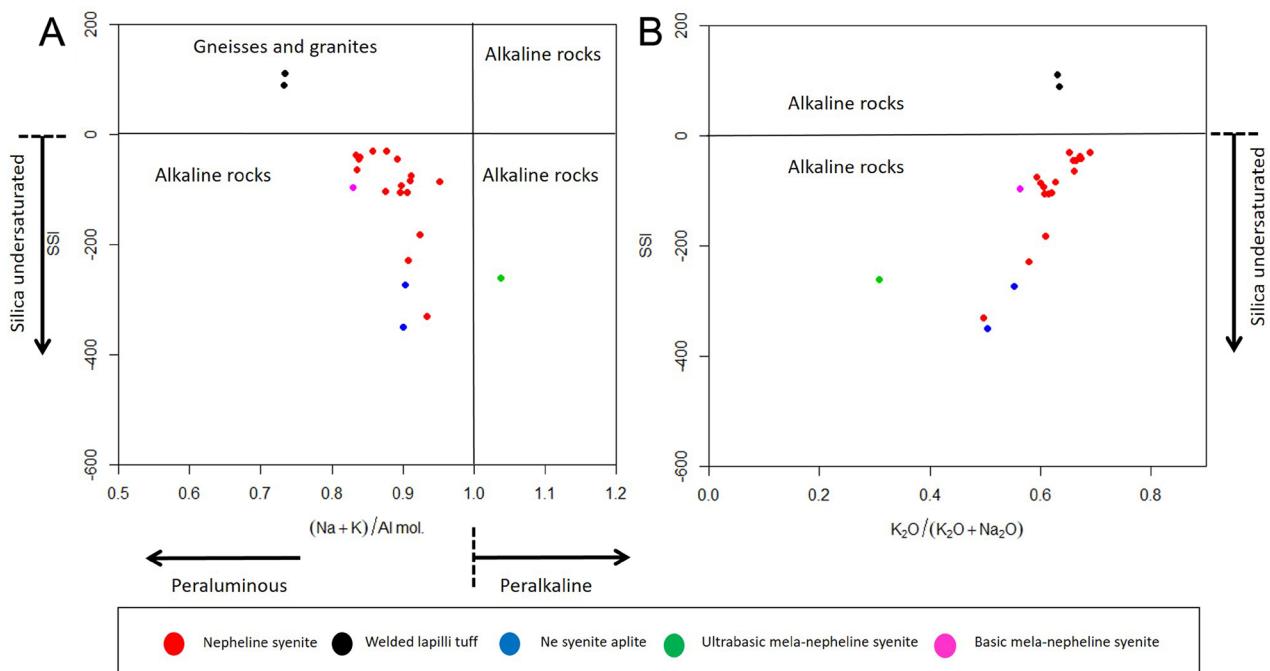


MSWD: mean squared weighted deviation.

**Figure 13.** U-Pb LA-ICP-MS dating diagrams for nepheline syenite samples of the Rio Bonito Complex. (A) Concordia diagram constructed with 13 analyses from the sample RB-01; (B) Concordia diagram constructed with 12 analyses from the sample RB-02.



**Figure 14.** (A) Diagram  $^{176}\text{Hf}/^{177}\text{Hf}$  vs age for 24 analyses from sample RB-01; (B) diagram  $\epsilon_{\text{Hf}}$  vs age for 24 analyses from sample RB-01; (C) diagram  $^{176}\text{Hf}/^{177}\text{Hf}$  vs age for 20 analyses from sample RB-02; (D) diagram  $\epsilon_{\text{Hf}}$  vs age for 20 analyses from sample RB-02.



**Figure 15.** Variation diagrams related to the Silica Saturation Index (SSI). (A)  $(\text{Na} + \text{K})/\text{Al}$  mol. vs. SSI; (B)  $\text{K}_2\text{O}/(\text{K}_2\text{O} + \text{Na}_2\text{O})$  vs. SSI.

normative quartz of 8.28, 6.70, 5.58, and 4.48 wt%. The alkali feldspar syenites of Tanguá and Cabo Frio Island have minor amount of normative quartz, 0.23 and 0.83 wt%. The felsic rocks of Soarinho massif have normative quartz. They plot on the area of non-alkaline rocks due to strong effects of continental crust assimilation. Among them, trachytic rocks have normative corundum.

Primary alkaline magmas are derived, in general, from water deficiency and Low-degree partial melting of an upper mantle source (McKenzie and Bickle, 1988). Bonin (1990) stated that during magma ascend and solidification, water content from wall rocks and/or after fractional crystallization may play an important role in the melt changing its composition moving from under-saturated to over-saturated alkaline melts.

The continuous geochemical trends between incompatible compositions can be formed by magma super-reheating (Motoki et al., 2010, 2013; Sichel et al., 2012). The super reheated nepheline syenite magma melts the country rock generating  $\text{SiO}_2$ -oversaturated magma. Because of the hyper-liquidus temperature, the  $\text{SiO}_2$ -undersaturated and oversaturated liquids can mix generating thermodynamically unstable alkali feldspar syenite magma. If the country rock melting is only partial, the  $\text{SiO}_2$ -oversaturated liquid should be pegmatitic with peraluminous composition. The volatile components extracted from the country rocks decrease the liquidus temperature, which makes easier the mixture between the liquids with incompatible compositions.

The nepheline syenite aplite has low total REEs, concave REE pattern, and positive Eu anomaly. These observations indicate that the aplite magma is not derived directly from the nepheline syenite magma of the main intrusive body. In addition, this rock has no cumulative plagioclase, and therefore, pointing to a reducing environment.

The large range of negative epsilon Hf can elucidate a crustal provenance for this alkaline massif contrasting with previous work based on a mantle plume origin (Thompson et al., 1998 and references therein). Both analyzed samples (RB-01 and RB-02) of Rio Bonito massif have large variation in initial  $^{176}\text{Hf}/^{177}\text{Hf}$  ratios (see figures above). Griffin et al. (2002) and Andersen et al. (2002) pointed that intra sample isotopic range more likely reflects Hf isotope variations in the magma during igneous zircon growth. According to Ketchum et al. (2008), such variation is due to one or more processes that can occur in the magma source region or magma chamber. Zircon crystallization during magma mixing episodes can yield both inter- and intragrain variations in Hf isotope composition. Magma interaction with isotopically dissimilar wall rocks may also generate isotopic heterogeneity during zircon crystallization (Ketchum et al., 2008). A wide spread of  $\epsilon_{\text{Hf(t)}}$  within a single zircon crystal or among zircons from a single rock sample or thin section may result from magma

mixing or contamination (Griffin et al., 2002; Kemp et al., 2007; Shaw and Flood, 2009) or indicate preservation of source heterogeneity (Farina et al., 2014).

Published Sr and Nd initial isotopic ratios (Gibson et al., 1995, 1997; Thompson et al., 1998; Ulbrich et al., 2003, 2005) point to a lithospheric mantle source for the southeastern Brazil alkaline rocks. On the other hand, our data show a set of strongly negative (-6 to -33)  $\epsilon_{\text{Hf}}$ , indicating a possible crustal source corroborated by the SSI. Thus, the SSI could indicate not only crustal assimilation but also crustal genesis. The large variations in  $^{176}\text{Hf}/^{177}\text{Hf}$  ratio showed by the Rio Bonito samples indicate zircons crystallizing from a magma undergoing assimilation of older crust, which is corroborated by trace element ratios (Rb/Ba and Rb/Sr).

## CONCLUSIONS

The geochemical, geochronological, and isotopic data of the alkaline rocks of Rio Bonito intrusive complex, State of Rio de Janeiro, Brazil, lead to the following conclusions:

- Most of the analysed rocks are  $\text{SiO}_2$ -undersaturated and meta-alkaline with relatively low  $(\text{Na} + \text{K})/\text{Al}$  and  $\text{K}_2\text{O}/(\text{Na}_2\text{O} + \text{K}_2\text{O})$ , being classified as potassic nepheline syenite. The magma fractionation decreases  $\text{K}_2\text{O}/(\text{Na}_2\text{O} + \text{K}_2\text{O})$ , increases  $(\text{Na} + \text{K})/\text{Al}$ ,  $\text{Rb}/\text{K}$ , and  $\text{Zr}/\text{TiO}_2$ , and the residual magma becomes more  $\text{SiO}_2$ -undersaturated.
- The variation diagrams related to  $\text{TiO}_2$ ,  $\text{Fe}_2\text{O}_3^*$ ,  $\text{MgO}$ ,  $\text{CaO}$ , and  $\text{P}_2\text{O}_5$  indicate in-situ crystallisation and fractionation of titanite, ilmenite, apatite, and clinopyroxene or amphibole. The less fractionated nepheline syenite has unexpectedly low  $\text{Zr}/\text{TiO}_2$ .
- The RRE pattern is linear with negative gradient. Nepheline syenite aplite has low total REEs and concave pattern with positive Eu anomaly.
- The ultrabasic and basic mela-nepheline syenite samples have total REEs and light REEs concentration higher than those of the felsic alkaline rocks. These observations indicate that the nepheline syenite magma is not derived directly from the alkaline ultramafic magma by fractional crystallisation.
- Normalized REE diagram indicate a strong amphibole fractionation signature based on concavity upward with retention of middle REE.
- The laser-spot step-heating  $^{40}\text{Ar}/^{39}\text{Ar}$  age for biotite is  $65.03 \pm 0.70$  Ma and that for the amphibole is  $65.03 \pm 0.46$  Ma. The U-Pb ages ( $65.49 \pm 0.30$  and  $65.18 \pm 0.30$  Ma) shows that the crystallization age is indistinguishable of the cooling age. Those ages are close to the whole rock Rb-Sr age of adjacent intrusive complex of Tanguá.
- Lu-Hf data from two samples of nepheline syenite shows average  $\epsilon_{\text{Hf}}$  values of -7 to -33, indicating that there is an

important crustal contribution to the formation of nepheline syenite magma.

## ACKNOWLEDGEMENT

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**Appendix 1.** U-Pb data of 28 analyses from the sample RB-01.

Spot	f206	$^{207}\text{Pb}$	$^{206}\text{Pb}$	U	Th/U	$^{207}\text{Pb}'$	$^{206}\text{Pb}'$	2 s (%)	$^{207}\text{Pb}'$	$^{206}\text{Pb}'$	2 s (%)	$^{206}\text{Pb}'$	$^{207}\text{Pb}'$	$^{206}\text{Pb}'$	2 s (%)	% conc									
7	0,1723	244,022	11,525	1011	0,75	0,04722349	1,30	0,069706341	3,79	0,010704514	3,56	0,94	61	31,05	69	2,42	68	2,50	100						
8	0,8516	124,819	5,856	543	0,57	0,046917016	4,05	0,065998658	17,34	0,010202426	16,86	0,97	45	96,74	65	10,94	65	10,91	101						
9	0	22,033	1,042	97	0,98	0,0473	3,28	0,066013489	3,94	0,010122092	2,18	0,55	64	78,03	65	1,40	65	2,47	100						
10	1	159,985	7,600	670	0,68	0,04750387	4,36	0,069389293	19,51	0,010594054	19,02	0,97	75	103,65	68	12,81	68	12,89	100						
11	0,2643	290,787	13,727	1250	0,56	0,047204909	1,56	0,067187936	5,59	0,010322927	5,37	0,96	60	37,09	66	3,52	66	3,57	100						
12	0	107,956	5,111	464	0,62	0,04734	1,31	0,067314669	2,04	0,010312885	1,56	0,77	66	31,18	66	1,02	66	1,30	100						
13	1	153,620	7,285	652	1,83	0,047420077	4,43	0,068347715	19,78	0,01045347	19,27	0,97	70	105,49	67	12,81	67	12,88	100						
14	0	106,784	5,049	468	0,44	0,04728	1,71	0,065985577	2,47	0,010122092	1,79	0,72	63	40,80	65	1,15	65	1,55	100						
15	0	227,234	10,753	966	1,44	0,04732	1,10	0,068072437	1,89	0,010433386	1,54	0,81	65	26,16	67	1,02	67	1,22	100						
16	0	62,408	2,951	272	0,59	0,04729	1,61	0,066261436	2,40	0,010162259	1,78	0,74	64	38,27	65	1,15	65	1,51	100						
17	0	81,140	3,839	350	0,36	0,04731	1,31	0,067099997	2,04	0,010272718	1,56	0,77	65	31,20	66	1,02	66	1,30	100						
18	1	101,181	4,796	442	0,61	0,047398616	4,42	0,066348	20,33	0,010152217	19,85	0,98	69	105,25	65	12,82	65	12,88	100						
19	0	51,421	2,541	228	0,63	0,04942	1,50	0,06821956	2,19	0,010011633	1,60	0,73	68	34,98	64	1,02	67	1,42	96						
20	0	74,529	3,527	323	0,51	0,04732	1,44	0,066996575	2,13	0,010222509	1,57	0,74	65	34,21	66	1,02	66	1,35	100						
27	0,8335	41,061	1,927	181	0,43	0,046935524	4,30	0,064984934	17,33	0,010041758	16,79	0,97	46	102,81	64	10,72	64	10,75	101						
28	0,5223	190,109	8,951	824	1,8	0,047082817	2,53	0,066492648	10,66	0,010242593	10,36	0,97	53	60,31	66	6,75	65	6,75	100						
29	1,0588	145,851	6,839	642	1,19	0,046888227	4,90	0,065179127	21,73	0,010081925	21,17	0,97	44	117,12	65	13,58	64	13,54	101						
30	4,3958	54,446	2,462	239	1,82	0,046211233	19,53	0,062910542	89,65	0,0100994967	87,49	0,98	-44	474,65	65	56,37	62	55,13	104						
31	0	40,836	1,932	180	1,08	0,0473	2,47	0,065686042	3,18	0,010071883	1,99	0,63	64	58,90	65	1,28	65	1,98	100						
32	0,4719	162,776	7,699	630	1,04	0,047295775	2,35	0,074716928	8,74	0,011457646	8,42	0,96	64	56,04	73	6,13	73	6,17	100						
33	0,1182	145,033	6,852	633	0,59	0,047244079	1,29	0,066262505	3,10	0,010172301	2,82	0,91	62	30,66	65	1,82	65	1,95	100						
34	0,3496	203,810	9,609	895	0,26	0,047144588	2,04	0,065666043	7,42	0,010102008	7,13	0,96	57	48,64	65	4,58	65	4,63	100						
35	0	111,799	5,294	485	0,98	0,04735	1,22	0,066738859	1,99	0,010222509	1,57	0,79	67	29,16	66	1,02	66	1,26	100						
36	0	155,449	7,354	678	0,98	0,04731	1,14	0,066354963	1,95	0,010172301	1,58	0,81	65	27,18	65	1,02	65	1,23	100						
37	0	157,705	7,466	683	0,36	0,04734	1,14	0,066855854	1,94	0,010242593	1,57	0,81	66	27,15	66	1,02	66	1,23	100						
38	0	67,939	3,216	297	1,91	0,04733	1,84	0,066251952	2,56	0,010152217	1,78	0,70	66	43,76	65	1,15	65	1,61	100						
39	0,2266	82,632	3,902	360	0,23	0,047222747	1,98	0,066363351	5,20	0,010192384	4,80	0,92	61	47,28	65	3,11	65	3,28	100						
40	1,882	113,837	5,290	496	1,38	0,046468706	8,27	0,065663678	38,03	0,010192384	37,12	0,98	22	99,56	65	24,99	64	23,85	102						

**Appendix 2.** U-Pb data of 28 analyses from the sample RB-02.

Spot	f206	<sup>207</sup> Pb'	<sup>206</sup> Pb	U	Th/U	<sup>207</sup> Pb'	<sup>206</sup> Pb'	2 s (%)	<sup>207</sup> Pb'	<sup>235</sup> U	2 s (%)	<sup>206</sup> Pb'	<sup>238</sup> U	2 s (%)	Rho	<sup>207</sup> Pb'	<sup>206</sup> Pb'	2 s	<sup>207</sup> Pb'	<sup>235</sup> U	2 s	% conc
47	1,8508	18,579	863	80	0,94	0,046473623	8,71	0,067397389	36,38	0,010518051	35,32	0,97	22	209,11	67	23,70	66	23,55	102			
48	0	47,425	2,592	206	1,48	0,05465	2,36	0,07880212	3,04	0,010457948	1,92	0,63	398	52,94	67	1,28	77	2,25	87			
49	0	69,814	3,305	309	2,31	0,04734	1,42	0,06675768	2,11	0,010227552	1,57	0,74	66	33,69	66	1,02	66	1,34	100			
50	0	35,359	1,674	160	1,6	0,04733	1,77	0,06524005	2,39	0,009997157	1,60	0,67	66	42,25	64	1,02	64	1,49	100			
51	0	66,555	3,149	302	2,53	0,04731	1,67	0,065212482	2,31	0,009997157	1,60	0,69	65	39,76	64	1,02	64	1,44	100			
52	0	37,349	1,768	164	1,52	0,04733	2,24	0,067201174	2,84	0,010297673	1,75	0,62	66	53,32	66	1,15	66	1,82	100			
53	0,4157	117,753	5,548	527	3,08	0,047113316	2,18	0,0657873	8,65	0,01012738	8,38	0,97	55	51,92	65	5,41	65	5,43	100			
54	1	56,389	2,673	246	1,85	0,047395148	4,56	0,068013742	19,86	0,010407862	19,33	0,97	69	108,51	67	12,82	67	12,90	100			
55	1	136,382	6,476	610	2,21	0,047484426	4,40	0,066371089	20,31	0,010137393	19,83	0,98	74	104,53	65	12,82	65	12,89	100			
56	0,2593	147,266	6,961	638	3,01	0,047267102	1,65	0,068221695	5,45	0,010467965	5,19	0,95	63	39,21	67	3,46	67	3,53	100			
57	0,188	65,349	3,088	292	1,96	0,04725099	1,70	0,065979544	4,39	0,01012738	4,04	0,92	62	40,51	65	2,61	65	2,76	100			
58	0	60,197	2,850	265	1,9	0,04735	1,52	0,067098774	2,32	0,010277638	1,75	0,76	67	36,19	66	1,15	66	1,48	100			
59	0,9236	20,125	944	91	0,99	0,046912683	4,59	0,065183172	19,01	0,010077294	18,45	0,97	45	109,79	65	11,86	64	11,86	101			
60	0	95,723	4,532	423	3,62	0,04735	1,35	0,067033375	2,06	0,010267621	1,56	0,76	67	32,17	66	1,02	66	1,32	100			
67	0	93,892	4,444	417	3,4	0,04733	1,35	0,066678207	2,07	0,010217535	1,57	0,76	66	32,19	66	1,02	66	1,31	100			
68	0,1539	91,476	4,327	400	2,92	0,047297106	1,67	0,067677073	3,83	0,01037781	3,44	0,90	64	39,89	67	2,28	66	2,46	100			
69	0	124,010	5,873	549	2,42	0,04736	1,27	0,066916708	2,01	0,010247587	1,56	0,78	67	30,15	66	1,02	66	1,28	100			
70	1	121,800	5,772	537	2,6	0,047387837	4,59	0,067152392	20,10	0,010277638	19,57	0,97	69	109,31	66	12,83	66	12,91	100			
71	2,2069	72,085	3,336	316	2,54	0,046285474	9,69	0,065973824	43,89	0,010337741	42,80	0,98	13	233,05	66	28,25	65	27,92	102			
72	0,4917	90,269	4,251	399	2,77	0,047097257	2,50	0,066545468	10,06	0,010247587	9,74	0,97	54	59,72	66	6,36	65	6,38	100			
73	0,2946	161,794	7,635	715	4,89	0,047190544	1,75	0,066742455	6,22	0,010257604	5,96	0,96	59	41,77	66	3,90	66	3,95	100			
74	0	100,258	4,747	444	2,2	0,04735	1,46	0,06683718	2,14	0,010237569	1,57	0,73	67	34,68	66	1,02	66	1,36	100			
75	1	40,737	1,945	185	1,43	0,047748398	4,71	0,065618927	20,72	0,009967105	20,18	0,97	87	111,72	64	12,83	65	13,02	99			
76	0	187,948	8,901	834	4,84	0,04736	1,20	0,066720471	1,98	0,010217535	1,57	0,79	67	28,64	66	1,02	66	1,25	100			
77	0	68,003	3,218	306	2,08	0,04732	1,42	0,065618408	2,13	0,01005726	1,59	0,75	65	33,71	65	1,02	65	1,33	100			
78	1,5436	68,796	3,204	315	2,19	0,046579733	6,88	0,063626865	32,01	0,009907002	31,26	0,98	28	164,97	64	19,76	63	19,60	101			
79	0	79,183	3,745	363	2,25	0,0473	1,48	0,064545404	2,19	0,009896985	1,62	0,74	64	35,24	63	1,02	64	1,35	100			
80	0,3488	79,945	3,768	364	2,05	0,047134998	2,16	0,064775953	7,51	0,009967105	7,19	0,96	56	51,46	64	4,57	64	4,64	100			

**Appendix 3.** Lu-Hf data of 23 analyses from the sample RB-01.

Spot	$^{176}\text{Yb}/^{177}\text{Hf}$	$\pm 2\text{s}$	$a$	$^{176}\text{Lu}/^{177}\text{Hf}$	$\pm 2\text{s}$	$b$	$^{178}\text{Hf}/^{177}\text{Hf}$	$\pm 2\text{s}$	$c$	$^{176}\text{Hf}/^{177}\text{Hf}$	$\pm 2\text{s}$	$d$	$\varepsilon_{\text{Hf}(t)}$	$\pm 2\text{s}$	$e$	Avg MORB	TDM2 f	TDM2 g	age	$\pm 2\text{s}$
22	0,1116	90	0,00288	18	1,46671	1,88707	6	0,2824246	22	0,282422043	-11,39395	1,22168	1,4	1,7	1,4	65	0			
23	0,244	210	0,00597	41	1,46669	1,88676	7	0,282102	21	0,282094551	-22,97656	2,24845	2	2,3	2	65	0			
24	0,1206	144	0,00274	29	1,46671	1,88681	8	0,28238	25	0,282376223	-13,03666	1,65981	1,5	1,8	1,5	65	0			
25	0,0866	87	0,00219	18	1,46669	1,88687	7	0,282436	24	0,282433095	-11,00305	1,19976	1,4	1,6	1,4	65	0			
26	0,0582	47	0,00133	8	1,46671	1,88684	9	0,28251	30	0,282508308	-8,34296	0,94779	1,3	1,5	1,2	65	0			
27	0,0769	63	0,00176	11	1,46663	1,88685	8	0,282464	28	0,282461597	-9,97281	1,03728	1,3	1,6	1,3	65	0			
28	0,162	163	0,0039	29	1,46664	1,88683	7	0,282255	21	0,282250177	-17,47244	1,82229	1,8	2	1,7	65	0			
29	0,1197	98	0,00254	17	1,46666	1,88704	8	0,282352	27	0,282349281	-13,96735	1,27626	1,6	1,8	1,5	65	0			
30	0,1321	118	0,00313	21	1,46667	1,88683	6	0,282334	25	0,282329876	-14,65366	1,43728	1,6	1,8	1,6	65	0			
31	0,0701	59	0,00162	11	1,46667	1,88699	7	0,282489	30	0,282486919	-9,07722	1,01416	1,3	1,5	1,3	65	0			
32	0,0837	70	0,00196	12	1,46671	1,88692	8	0,282464	22	0,28246121	-10,00868	1,0836	1,3	1,6	1,3	65	0			
33	0,0658	56	0,00156	10	1,46671	1,88708	7	0,282494	30	0,28249239	-8,92813	0,99771	1,3	1,5	1,3	65	0			
34	0,1501	134	0,00327	21	1,46665	1,8871	9	0,282285	20	0,282281297	-16,37179	1,56775	1,7	1,9	1,7	65	0			
35	0,1267	107	0,00295	18	1,46669	1,88685	7	0,282368	22	0,282364063	-13,42237	1,35257	1,5	1,8	1,5	65	0			
36	0,1187	98	0,00262	18	1,46662	1,88684	8	0,282376	29	0,282372303	-13,13095	1,28041	1,5	1,8	1,5	65	0			
37	0,1217	101	0,00264	18	1,4666	1,88691	8	0,28237	27	0,28236624	-13,30101	1,30542	1,5	1,8	1,5	65	0			
38	0,1279	120	0,00297	20	1,46659	1,88679	7	0,282381	27	0,28237712	-13,00493	1,45135	1,5	1,8	1,5	65	0			
39	0,2536	231	0,00586	48	1,46661	1,88696	8	0,282022	23	0,282014913	-25,77101	2,44777	2,2	2,4	2,2	65	0			
40	0,0791	89	0,0017	16	1,46655	1,88696	9	0,282495	23	0,282493355	-8,82741	1,21327	1,3	1,5	1,3	65	0			
41	0,0271	24	0,00063	4	1,46645	1,88683	7	0,282586	29	0,282585584	-5,60986	0,85845	1,1	1,4	1,1	65	0			
42	0,0574	46	0,00139	8	1,46646	1,88696	8	0,282518	28	0,282516121	-8,02223	0,94497	1,2	1,5	1,2	65	0			
43	0,1234	101	0,00277	17	1,46645	1,88708	9	0,282379	23	0,282375215	-13,05014	1,30355	1,5	1,8	1,5	65	0			
44	0,0874	71	0,00198	12	1,46651	1,88701	8	0,282485	32	0,282482023	-9,18381	1,08881	1,3	1,6	1,3	65	0			

**Appendix 4.** Lu-Hf data of 20 analyses from the sample RB-02.

Spot	$^{176}\text{Yb}/^{177}\text{Hf}$	$\pm 2\text{s}$	$a$	$^{176}\text{Lu}/^{177}\text{Hf}$	$\pm 2\text{s}$	$b$	$^{178}\text{Lu}/^{177}\text{Hf}$	$\pm 2\text{s}$	$c$	$^{176}\text{Hf}/^{177}\text{Hf}$	$\pm 2\text{s}$	$d$	$\varepsilon_{\text{Hf}(t)}$	$\pm 2\text{s}$	$e$	Avg MORB	TDM2 f	TDM2 g	age	$\pm 2\text{s}$
126	0,2969	239	0,00894	54	1,46695	1,88679	5	0,281816	25	0,281805346	-33,18292	2,51925	2,6	2,8	2,6	65	0			
127	0,2375	261	0,00552	63	1,46685	1,88697	7	0,282033	17	0,282026521	-25,36048	2,73426	2,2	2,4	2,2	65	0			
128	0,0827	67	0,0021	14	1,46679	1,88678	6	0,28246	16	0,282457171	-10,12933	1,06324	1,4	1,6	1,3	65	0			
129	0,1431	117	0,00366	22	1,4668	1,88683	6	0,282319	13	0,282314227	-15,20715	1,42275	1,6	1,9	1,6	65	0			
130	0,1243	108	0,0032	22	1,46689	1,88708	6	0,282384	17	0,282379584	-12,87342	1,35514	1,5	1,8	1,5	65	0			
131	0,1093	91	0,00262	17	1,4668	1,88701	7	0,282436	16	0,282432603	-11,04264	1,22402	1,4	1,6	1,4	65	0			
132	0,1596	141	0,00374	25	1,46686	1,88688	6	0,282276	17	0,282271298	-16,68109	1,63464	1,7	2	1,7	65	0			
133	0,1749	140	0,0042	25	1,4668	1,88689	6	0,282246	15	0,282241167	-17,76891	1,6255	1,8	2	1,8	65	0			
134	0,1663	133	0,00399	24	1,46672	1,88676	6	0,282271	20	0,282266234	-16,90452	1,56642	1,7	2	1,7	65	0			
135	0,1098	88	0,00274	16	1,46675	1,88689	6	0,28239	18	0,282386174	-12,64035	1,20566	1,5	1,7	1,5	65	0			
136	0,0573	46	0,00142	9	1,46668	1,88685	6	0,282519	21	0,28251767	-8,01183	0,94417	1,2	1,5	1,2	65	0			
137	0,0574	46	0,00151	9	1,46671	1,88707	6	0,28254	21	0,282537738	-7,27989	0,94382	1,2	1,4	1,2	65	0			
138	0,1345	108	0,00323	19	1,46679	1,88699	5	0,282364	21	0,282359984	-13,57169	1,35492	1,5	1,8	1,5	65	0			
139	0,208	166	0,00496	30	1,46677	1,88699	6	0,282232	21	0,282226356	-18,31494	1,85483	1,8	2	1,8	65	0			
140	0,0602	48	0,00155	9	1,46672	1,88681	6	0,282526	26	0,282524107	-7,73979	0,95503	1,2	1,5	1,2	65	0			
141	0,0562	45	0,00133	8	1,4668	1,88678	8	0,282512	16	0,282510729	-8,23512	0,93895	1,3	1,5	1,2	65	0			
142	0,1755	140	0,00429	26	1,46675	1,88684	6	0,282226	16	0,282220829	-18,48825	1,62545	1,8	2,1	1,8	65	0			
143	0,1321	106	0,00313	19	1,46667	1,88683	6	0,282342	16	0,282337824	-14,35039	1,3393	1,6	1,8	1,6	65	0			
144	0,0557	45	0,00139	8	1,46686	1,88677	7	0,282559	14	0,282557422	-6,58369	0,93732	1,2	1,4	1,1	65	0			
145	0,1217	97	0,00294	18	1,46675	1,88669	6	0,282375	15	0,282371869	-13,14627	1,27466	1,5	1,8	1,5	65	0			